



Curriculum Plan year 11 chemistry triple curriculum
2016

Autumn Term	Spring Term	Summer Term
<p>Controlled Assessment Completion</p> <p>Atomic Structure</p> <ul style="list-style-type: none"> • Atoms, Elements and Compounds • Isotopes/Electronic structure/Configurations • Balanced Equations • Ions and Ion formation • Separating Mixtures <p>The Periodic Table</p> <ul style="list-style-type: none"> • The Periodic Table and Development • Metals/Properties of Metals and Alloys • Transition Metals • Group 1/7/0 <p>Bonding, Structure and the properties of matter</p> <ul style="list-style-type: none"> • Ionic, covalent and metallic bonding • Relating Bonding and Structure to properties • Metals as conductors • States of Matter/State symbols • Properties of ionic and covalent molecules • Giant covalent structures (Diamond, graphite & Buckminsterfullerene) • Polymers and nanoparticles 	<p>Reactions of acids</p> <ul style="list-style-type: none"> • The pH scale and neutralisation • Reactivity Series and Metal Reactions • Reactions of metals with acids • Neutralisation of acids and Making salts • Soluble and Insoluble salts (Salts from Metals, Metal oxides and Hydroxides and Ammonia) <p>Electrolysis</p> <ul style="list-style-type: none"> • Process of electrolysis • Electrolysis of molten ionic compounds • Electrolysis for metal extraction • Electrolysis of Aluminium Ore • Electrolysis of aqueous solutions (CuSO₄ and NaOH) • Electroplating • Purifying Copper • Electrochemical Cells • Electrolysis half-equations 	<p>Analytical Techniques</p> <ul style="list-style-type: none"> • Testing for Gases (Cl₂, O₂, CO₂ and H₂) • Qualitative Analysis - Flame tests and NaOH test • Qualitative Analysis - Anions • Chromatography/ Instrumental methods of analysis <p>Quantitative Chemistry</p> <ul style="list-style-type: none"> • RAM and RFM • Balancing Equations • Empirical Formula • Predicting Masses • Reacting Masses • Moles and Molarity • Titration Calculations • Hard water • Atom Economy and Percentage Yield
<p>HALF TERM</p>		

Bishop Milner Catholic College



<p>Chemistry of the Atmosphere</p> <ul style="list-style-type: none">• Development of Atmosphere• How the Atmosphere changed• Greenhouse gases and global warming• Atmospheric pollutants and their effects <p>Crude Oil and Fuels</p> <ul style="list-style-type: none">• Fraction distillation of Crude Oil• Properties and Uses of Fractions• Environmental problems of Crude Oil/Alternative fuels• Cracking• Plant Oils and Uses• Unsaturated Oils/emulsions	<p>Energy Changes and Rates of Change</p> <ul style="list-style-type: none">• Monitoring Chemical Reactions• Exothermic and Endothermic reactions• Collision Theory• Factors affecting rate of reaction• Reversible Reactions• Haber process and Fertilizers• Ideal Gas Law• Avogadro's Law	<p>Organic Chemistry</p> <ul style="list-style-type: none">• Homologous Series• Alcohols• Carboxylic Acids• Esters• Soaps and Detergents <p>Revision for Summer Examinations</p>
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